

### Article Information

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#### Introduction

Ternary semiconductors are gaining ground as materials of choice in optoelectronic. Some of these materials contain toxic components examples are cadmium (Cd) and lead (Pb) based cells. A typical Pb based material is the PbSnS<sub>3</sub>, which have also attracted both experimental (Paar *et al.*, 2000; Kuku and Azi, 1998; Kuku *et al.*, 2006) and theoretical investigation (Omehe *et al.*, 2013). The Cu-Sn-S system is an alternative with prospect (Wu et al, 1986).

Materials in the system includes Cu<sub>2</sub>Sn<sub>3</sub>S<sub>3</sub>, Cu<sub>2</sub>SnS<sub>3</sub>, Cu<sub>4</sub>SnS<sub>4</sub> and Cu<sub>10</sub>Sn<sub>2</sub>S<sub>13</sub>. In this study, we shall investigate the electronic band structure Copper Tin Sulphide Cu<sub>4</sub>SnS<sub>4</sub> (CTS). It has attracted its own fair share of experimental and theoretical studies. Cu<sub>2</sub>SnS<sub>3</sub> was synthesized by Wu et al. (2007) and the experimental results did give the unique metallic character of hexagonal Cu<sub>2</sub>SnS<sub>3</sub>. Avellaneda et al. (2010) deposited thin films of copper sulphide (CuS, 200nm thick) over thin films of Tin Sulphide (SnS, 180nm thick) by sequential chemical deposition. The layers were heated in nitrogen atmosphere at 350°c and 400°c. The grazing incidence x-ray diffraction analysis revealed ternary composition, Cu<sub>2</sub>SnS<sub>3</sub> and Cu<sub>4</sub>SnS<sub>4</sub>. The optical band gaps of the films are direct, 0.95eV for Cu<sub>2</sub>SnS<sub>3</sub> and 1.2 eV for Cu<sub>4</sub>SnS<sub>4</sub>. Fernandes (2010) grew CTS ternary chalcogenide compounds by sulphurization of demagnetron sputtered metallic precursors. Cu<sub>2</sub>SnS<sub>3</sub> was formed at a maximum sulphurization temperature of 350°c while Cu<sub>4</sub>SnS<sub>4</sub> was obtained at a sulphurization temperature of about 400°c. The estimated band gap values were between 1.0 eV to 1.6 eV. Using the direct-coating from single metal organic precursor solution, Tiweri et al. (2013) deposited Cu2SnS3 thin films with a band gap of about 1.12 eV. Guan et al. (2013) deposited Cu<sub>2</sub>SnS<sub>3</sub> and Cu<sub>4</sub>SnS<sub>4</sub> thin films by successive ionic layer absorption and reaction method. A band gap value of 1.05 eV was obtained for tetragonal Cu<sub>2</sub>SnS<sub>3</sub>. Anuar et al. (2008) investigated the effect of varying the electrolytes concentration on the film's properties. The resulting substrate was analysed using X-ray diffraction technique. Their results show CTS to be p-type semiconductor band gap energy of 1.7 eV. In a related study (Anuar et al., 2010), they studied the effect of deposition time on the crystal's properties. They reported a band gap value of about 1.6-2.1 eV. The co-evaporation method was used by Vani et al. (2013) to study the effect of temperature variation on the properties of CTS. The Phillips' X-ray diffractometer was used in analyzing the structural properties of the film. Band gap values of about 1.7 to 1.94 eV were observed.

The electronic structure of orthorhombic CTS was investigated by Goto *et al.* (2013) within the framework of the density functional theory (DFT) using the projector augmented wave (PAW) method. For exchange and correlation, they used the pedew-Burke-Ernzerhof parametrization (PBE). Their calculated band gap value was 0.39 which fall short of the experiment value.

In this study, we intend to investigate the electronic band structure of  $Cu_4SnS_4$  using PAW in conjunction with the LDA+U scheme. To the best of our knowledge, this is the first time a theoretical investigation of the electronic band structure would be carried out on CTS.

# Methods

Cu<sub>4</sub>SnS<sub>4</sub> belongs to the orthorhombic structure with the space group Pnma (space group number 62). The lattice parameters and atomic coordinates shown in table 1 were adopted from (Goto *et al.*, 2013). The Wyckoff's atomic positions are shown in Table 1. The unit cell is made up of four formula unit, which gives a total of 36 atomic coordinates in seven (7) types. The structure of Cu<sub>4</sub>SnS<sub>4</sub> used in this calculation is shown in Figure 1.

The paw for Cu contains the 3s and 3p electrons as semi core. The band structure calculations utilized the LDA+U technique within the DFT framework as implemented in the abinit package (Gonze *et al.*, 2002; Gonze *et al.* 2005).

 Table 1: Lattice parameters and Wyckoff's atomic positions

a = 1.3608 nm b = 0.7600 nm and c = 0.6482 nm

a = 1.5008  mm, b = 0.7650  mm and $c = 0.0482  mm$			
Atom	X	Y	Z
Cu1	0.3341	0.9894	0.1182
Cu2	0.3375	0.25	0.5669
Cu3	0.4286	0.25	0.9017
Sn	0.083	0.25	0.1144
S1	0.1761	0.9964	0.2586
S2	0.0824	0.25	0.7288
<b>S</b> 3	0.4094	0.25	0.2514

#### Abstract

A Study of the Electronic Band Structure of Cu<sub>4</sub>SnS<sub>4</sub>

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The electronic band structure of thin film of the orthorhombic Cu<sub>4</sub>SnS<sub>4</sub> using the pseudopotential method within density functional theory framework was performed. LDA+U technique and the projector augmented wave PAW was used in all the calculations. The calculated results showed that Cu<sub>4</sub>SnS<sub>4</sub> is p-type semiconductor with a band gap energy value of 1.57 eV. It was also discovered that the gamma to X transition was critical in the structure of the material.



Figure 1: The structure of CTS used in this work. The blue balls represent Cu, the yellow balls represent S and the gray balls represent Sn.



Figure 2: The electronic band structure of CTS from -20 to 3 eV

A kinetic energy cutoff of 20 Ha was used for the generation of plane waves. For the k-points, a Monkhorst-Pack (1976) grid of 6x6x1 was used for Brillouin zone integration.

# Results

Figures 2, 3 and 4 displays the electronic band structure of CTL within the energy range of -85 to 3 eV along lines of high symmetry in the Brillouin zone. Figure 5 shows the total density of states (TDOS) with its Fermi energy level at 0.1 Ha while figures 6 to 12 display the partial density of states (PDOS) for CTS.



Figure 3: The electronic band structure of CTS from -45 to 0 eV



Figure 5 shows the total density of states, the band gap value is also well reproduced in the TDOS. Our calculated TDOS differ quantitatively and qualitatively from that of (Goto *et al.*, 2013).Figures 6 to 12 displays the partial density of state (PDOS) for CTS, these represent the contribution to TDOS from the various atomic species used in the computations. The Cu-3s and Cu-3p states contribution to TDOS can be seen about the energy

range of -3 to -2.8 Ha. These contributions are from the three Cu types as seen from figures 6 to 8. The S-3s dominates the states about the -1.5 Ha to -0.65 Ha energy range. The states immediately below the Fermi level are mainly Cu-3d, S-3p, Sn-5s and Sn-4d orbitals. The conduction band is predominantly contributions from Sn-5s and S-3p orbitals.



Figure 5: The total density of states for Cu<sub>4</sub>SnS<sub>4</sub>. The Fermi energy is 0.1 Ha.



Figure 6: The partial density of states for Cu(1). The Fermi energy is 0.1 Ha.



Figure 7: The partial density of states for Cu(2). The Fermi energy is 0.1 Ha.



Figure 8: The partial density of states for Cu(3). The Fermi energy is 0.1 Ha.



Figure 9: The partial density of states for S(1-3). The Fermi energy is 0.1 Ha.



Figure 10: The partial density of states for Sn. The Fermi energy is 0.1 Ha.

### Discussion

It can be seen that, the valence band maximum (VBM) and the conduction band minimum (CBM) are located at the gamma point resulting in a direct energy gap of 1.57 eV (Figures 2 - 4). The CBM occur at both the gamma and the x-points of high symmetry. This makes it a critical transition in the structure. It may also indicate the material's versatility, that is, it can be used for applications requiring direct and indirect band gaps. The direct band gap of 1.57 eV is in agreement with the experimental results. This result is an improvement over the LDA results of 0.39 eV obtained by Goto et al. (2013). This is as a result of the addition of the U term to the LDA. Gunnarsson and Schonhammer (1986) showed that the discontinuity in the one-electron potential can give a large contribution to the band gap. To overcome this deficiency, an orbital dependent correction is added to the LDA potential. This induces an upward shift of the unoccupied states and a downward shift of the occupied states. The screen coulomb interaction U and the exchange parameter J used in the calculations were taken from (Marel and Sawatzky, 1988).

# Conclusion

The electronic band structure of  $Cu_4SnS_4$  (CTS) has been carried out using the LDA+U technique in conjunction with the projector augmented wave (PAW). This work was done within the framework of the density functional theory (DFT). It was observed that CTS is an n-type semiconductor with a band gap energy value of 1.57 eV.

## References

Anuar, K., Ho, S.M., Tan, W. T., Atan, M. S., Kuang, D., Jelas, H. and Nagalingam, S. (2008). Effect of solution concentration on the properties of Cu<sub>4</sub>SnS<sub>4</sub> thin films, *Mater. Sci.* 14, No. 2,101-105

Anuar, K., Nagalingam, S., Tan, W. T. and Ho, S. M. (2010). Effect of deposition period on the chemical bath deposited Cu<sub>4</sub>SnS<sub>4</sub> thin films, *Rev Soc Quim Peru*. 76 (1), 55-60.

Avellaneda, D., Nair, M. T. S. and Nair, P. K. (2010).  $Cu_2SnS_3$ and  $Cu_4SnS_4$  thin film via chemical deposition for photovoltaic application, Journal of Electrochemical society, 157 (6), D346-D352

Fernanders, P. P. A. (2010). A study of ternary  $Cu_2SnS_3$  and  $Cu_3SnS_4$  thin films prepared by sulphurizing stacked metal precursors, Journal of Physics, 43, 215403-215412. DOI. http://dx.doi.org/10.1088/0022-3727/43/21/215403

Gaun, H., Hunglie, S., Chao, G. and Viancung, H. (2013). Structure and optical properties of Cu<sub>2</sub>SnS<sub>3</sub> and Cu<sub>4</sub>SnS<sub>4</sub> thin films by successive ionic layer adsorption and reaction, Journal of materials science, materials in electronics, 25 (5) 1490-1494.

Gonze, X., Rignanese, G.-M., Verstraete, M., Beuken, J.-M., Pouillon, Y., Caracas, R., Jollet, F., Torrent, M., Zerah, G., Mikami, M., Ghosez, Ph., Veithen, M., Raty, J.-Y., Olevano, V., Bruneval, F., Reining, L., Godby, R., Onida, G., Hamann, D. R., and Allan, D. C. (2005). A brief Introduction to the Abinit software package. Z. Kristallogr. 220, 558-562.

Gonze, X., Beuken, J.-M., Caracas, R., Detraux, F., Fuchs, M., Rignanese, G.-M., Sindic, L., Verstraete, M., Zerah, G., Jollet, F., Torrent, M., Roy, A., Mikami, M., Ghosez, P.H., Raty, J.Y., and Allan, D.C. (2002). First-principles computation of material properties: the Abinit software project, Computational Materials Science 25, 478-492

Goto, Y., Kamihara, Y. and Masanori, M. (2013). First principles calculations of electronic structure for orthorhombic and monoclinic Cu<sub>4</sub>SnS<sub>4</sub>, Phys. Status Solidi C 10, No. 7-8, 1127-1129

Gunnarsson, O. and Schonhammer, K. (1986). Density-functional treatment of an exactly solvable semiconductor model, Phy. Rev. Let. 56, 1968-1971

Kuku, T. A., Azi, S. O., Osasona, O. (2006). Electrical properties of vacuum evaporated PbSnS3 thin films, J Mater Sci 41, 1067-1071.

Kuku, T. A. and Azi, S. O. (1998). Optical properties of evaporated PbSn3 thin films, J Mater Sci 33, 3193-3196.

Marel, D. and Sawatzky, G. A. (1988). Electron-electron interaction and localization in d and f transition metals, Phys. Rev. B 37, 10674-10684.

Monkhorst, H. J. and Park, J. D. (1976). Special points for Brillouin zone integration, Phys. Rev. B 13, 5188-5192.

Omehe, N. N., Ehika, S. and Azi, S. O. (2013). Electronic and vibrational properties of PbSnS<sub>3</sub>, IOSR Journal of Electrical and Electronics Engineering, vol 5, issue 5, pp 12-17.

Paar, W. H., Miletich, R., Topa, D., Criddle, A. J., De Brodtkorb, M. K, Amthauer, G., and Tippelt, G. (2000). Suredaite, PbSnS3, a new mineral species, from the Pirquitas Ag-Sn deposit, NW-Argentina: mineralogy and crystal structure, American Mineralogist, 85, 1066-1075.

Tiweri, D., Chaudhuri, T. K., Shripathi, T., Deshpande, U. and Rauat, R. (2013). Nontoxic,earth abundant 2% efficient CTS solar cell based on tetragonal film direct coating from single metalorganic precursor solution, Solar energy materials and solar cells, vol. 113, pp 165-170.

Wu, D., Knowles, C. R. and Chang, L. L. Y. (1986). Copper-tin Sulphides in the system Cu-Sn-S, mineralogical magazine, vol. 50, 323-325.

Wu, C., Hu, Z., Wang, C., Shang, H., Yang, J and Xie, Y. (2007). Hexagonal Cu<sub>2</sub>SnS<sub>3</sub> with metallic character: Another category of conducting sulfides. Applied physics letters 91, S.143104 http://dx.doi.org/10.1063/1.2790491

Vani, G., Miles, R. W. and Reddy, R. (2013). Preparation and properties of Cu4SnS4 thin films, International Journal of Optoelectronic Engineering, 3(1), 1-5 DOI: 10.5923/j.ijoe. 20130301.01