

Keywords: automobile workshops, contamination, heavy metals, pollution, soil profile workshops in Sapere interopolis were investigated for neavy inetars and physico-chemical properties using standard methods. The pH ranged observed was 5.09 - 6.39. At all locations investigated, the heavy metals (Zn, Mn, Pb, Cu, Cr, and Cd) obtained were much higher than that from the control sample. This is an indication that automobile workshops situated within the metropolis are anthropogenic sources of heavy metals in the soils that requires attention. The concentrations of Zn, Cu and Cd at most locations and depths were also higher than the recommended critical limits. The concentrations of the metals generally decrease with increase in depths. The extent of contamination level based on the average metal contents obtained from soil samples at the different depths was 57.69 - 129.58 folds for Zn, Manganese was 27.74 - 69.20 fold, while Pb, Cu, and Cr, were respectively 17.95 - 55.97, 8.39 - 14.82, and 113 - 504 fold compared with the control soil. Cadmium was not detected in control soil but values ranging from 1.70 to 8.89 mg/kg were recorded for the automobile workshops soils. The degree of metal pollution was in order: Oleh Road > Reclamation/Green-Egbede Road > Awolowo/Okpe Road > Okerigwre Road > Akintola Road > Shell Road > Gana New Road > Ajogodo Road > Ogodo Road > McPherson Road.

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Introduction

Heavy metals are group of metals and metalloids with an atomic density greater than 5 gdm⁻³. They are widely distributed in the environment and are not biodegradable but can be transformed into different forms (species) with time as the organic molecules binding them decompose or soil conditions change (Alloway and Aryes, 1997).

Heavy metals occur naturally in rock-formation and ore minerals and so there is a range of normal background concentrations of these elements in soils, sediments, water and living organisms. The very low general level of their background concentration in the soils, and plants as well as their biological role, makes them micro or trace elements (Alloway and Aryes, 1997; Lacatusu, 1998). Some heavy metals such as Zn, Cu, Mg, and Cr are essential to maintain growth and normal functioning of living organism including Man. However, many may be toxic at higher concentrations as they tend to bioaccumulate in human bodies making them dangerous and therefore poses great health and environment risks (Igwe and Abia, 2006). Other elements such as Pb, Hg and Cd are not essential for metabolic activities and exhibit toxic properties even at very low concentrations (Klaassen, 2001; White *et al.*, 2007).

The anthropogenic sources of heavy metals include, waste from mining, metallurgical, chemical industries, battery manufacturing, dye and tannery, pigment manufacturing industries as well as Leachates from contaminated dump sites and petroleum activities (Biney et al., 1994). Heavy metals are also emitted from the burning of fossil fuel, incineration of wastes, resource recovery plants in relative high levels fly ash particles (Neal et al., 1990). One other source of heavy metal pollution in soil is that due to automobile workshops activities. Spent (waste) automobile oil contain oxidation products, sediments water and metallic particles resulting from machinery wears, organic and inorganic chemicals used in oil additives and metals that are present in fuel and transferred to the crankcase during combustion (EEA, 2007). These wastes including also wastes oils used for cleaning during servicing, metal scraps, used batteries etc., indiscriminately discarded on the soil by artisans, contaminates the soil. Percolation of Leachates from these materials poses threat to underground water and creates concern for the normal functioning of the ecosystem.

The objectives of this study are to investigate the profile of heavy metals in soils within automobile workshops in Sapele metropolis, and to ascertain the extent of metal pollution in these soils.

Materials and methods

Study area

The study area is within Sapele, located on the Fresh Water Forest of Delta State, Nigeria. Automobile workshops in the metropolis were chosen with ages of establishment over seven years, obtained by personal communication with artisans working on these sites. No waste management practice has been conducted in these workshops.

Samples collection and preparation

Composite soil samples were collected from ten (10) different mechanic workshops; at each workshop four different points were chosen using cluster random sampling technique to collect the sample. Four soil samples per point were collected at depths 0-15, 15-30, 30-45 and 45-60 cm using standard soil (hand) auger. Sample of uncontaminated soil were obtained in similar manner from nearby virgin land in the metropolis to serve as control. The geographical position coordinates of the sampled locations were identified and mapped using global position system (GPS) (Figure 1 and Table 1). The collected samples were transferred into a black polythene bag, properly labelled and transported to laboratory (Aller, 1989). The samples were air-dried for a period of two weeks in a well ventilated space (Boulding, 1994). A good representative was obtained using quartering method. The dried representative soil samples were crushed in porcelain mortal and sieved through 2 mm (10 mesh size) stainless sieve. The air-dried < 2mm soil samples were stored in airtight polythene bags and labelled prior to analysis. The aggregate sample from each location used for physico-chemical analysis was prepared by

mixing together equal amount of each of the samples from the four different depths.



Figure 1: Map of Sapele showing the sampled locations

Locations	Description	Co-ordinates
Α	Ogodo Road before Zik Secondary School	N05°53.996′ E005°40.290′
В	McPherson Road by Ikomi Junction	N05°53.746′ E005°40.129′
С	Reclamation Road by Green-Egbedi	N05°53.384′ E005°40.633′
D	Okpe Road by Awolowo	N05°53.384' E005°40.924'
E	Oleh Road close to Christ the King College	N05°53.204′ E005°40.971′
F	Akintola Road by Atufe Junction	N05°53.421′ E005°41.184′
G	Ajogodo Road opposite Christ Apostolic Church	N05°53.952′ E005°41.779′
Н	Along Gana-New Road	N05°54.701′ E005°45.025′
Ι	Mechanic village at Shell Road	N05°52.102′ E005°41.632′
J	Okirigwre by Granite depot	N05°52.102′ E005°42.528′
Х	Control-Virgin land at Ubeyiyi	N05°52.847' E005°42.528'

Table 1: Locations description and coordinates

Analysis of Sample

The physicochemical properties of the samples were determined according to the standard methods. Soil pH was determined in water according to the method of Folson *et al.*, (1981), Organic Carbon (by dichromate oxidation) as described by Nelson and Sommers (1982). Available phosphorus was estimated

following the procedures described by Olsen and Sommers (1982), while the ammonium acetate method was used to determine the Cation Exchange Capacity (CEC) as specified by Reeuwijk (1995). The particle size distribution (PSD) analysis of soil was carried out by the hydrometer method described by

Bouyoucos (1962). The total heavy metals in soils were determined with Atomic adsorption spectrophotometer (Model 210 VGP) after digestion with a mixture of Nitric and Perchloric acids (Tessier *et al.*, 1979). The analysis was carried out in triplicates and the results are expressed as mean and standard deviation from the mean (\pm) .

Statistical analysis

The analytical results were compiled to form a multi-element data base using excels as statistical software. Pearson's correlation coefficients and p-values were calculated for all possible variable pairs.

Results

Table 2 shows some physical chemical properties of the various automobile workshops aggregate soils samples along with the control. The mean concentrations of metals at each depth in each sampled location as well as the average metals concentrations at each depth are shown in Tables 3 and 5 respectively.

Table 2: Some physico-chemical properties of aggregate soil samples from automobile workshops in Sapele and the control (x) (dry weight)

Parameter	C /				-	Result ± S	D				
			Samples								
	Α	В	С	D	Ε	F	G	н	Ι	J	Х
рН	5.43 ± 0.02	5.71 ± 0.00	5.09 ± 0.01	5.40 ± 0.02	6.02 ± 0.00	5.43 ± 0.01	$\begin{array}{c} 5.60 \pm \\ 0.02 \end{array}$	6.10 ± 0.02	$\begin{array}{c} 6.39 \pm \\ 0.04 \end{array}$	$\begin{array}{c} 5.84 \pm \\ 0.00 \end{array}$	6.20 ± 0.02
Clay (%)	$\begin{array}{c} 9.90 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 8.90 \pm \\ 1.30 \end{array}$	$\begin{array}{c} 9.90 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 4.40 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 3.40 \pm \\ 0.90 \end{array}$	$\begin{array}{c} 1.40 \pm \\ 0.30 \end{array}$	$\begin{array}{c} 8.40 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 9.40 \pm \\ 0.90 \end{array}$	$\begin{array}{c} 4.90 \pm \\ 0.70 \end{array}$	$\begin{array}{c} 7.40 \pm \\ 0.00 \end{array}$	10.4 ± 1.00
Silt (%)	$\begin{array}{c} 10.00 \pm \\ 1.20 \end{array}$	$\begin{array}{c} 9.00 \pm \\ 1.80 \end{array}$	$\begin{array}{c} 10.00 \pm \\ 2.00 \end{array}$	$\begin{array}{c} 4.50 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 4.50 \pm \\ 0.90 \end{array}$	$\begin{array}{c} 5.50 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 6.50 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 5.50 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 9.00 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 10.00 \pm \\ 0.00 \end{array}$	8.50 ± 1.20
Sand (%)	$\begin{array}{c} 80.10 \pm \\ 1.10 \end{array}$	$\begin{array}{c} 82.10 \pm \\ 2.00 \end{array}$	$\begin{array}{c} 80.10 \pm \\ 1.50 \end{array}$	$\begin{array}{c} 91.10 \pm \\ 1.90 \end{array}$	$\begin{array}{c} 92.10 \pm \\ 2.00 \end{array}$	$\begin{array}{c} 93.10 \pm \\ 2.20 \end{array}$	$\begin{array}{c} 85.10 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 85.10 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 86.10 \pm \\ 1.60 \end{array}$	$\begin{array}{c} 82.60 \pm \\ 2.00 \end{array}$	81.1 ± 3.00
TOC (%)	$\begin{array}{c} 0.29 \pm \\ 0.40 \end{array}$	$\begin{array}{c} 0.16 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.58 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 0.45 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 0.58 \pm \\ 0.08 \end{array}$	$\begin{array}{c} 1.28 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.16 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.61 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 0.07 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.54 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 0.06 \pm \\ 0.01 \end{array}$
OM (%)	$\begin{array}{c} 0.50 \pm \\ 0.40 \end{array}$	$\begin{array}{c} 0.28 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 1.00 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 0.78 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 1.00 \pm \\ 0.08 \end{array}$	$\begin{array}{c} 2.21 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.28 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 1.06 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 0.12 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.93 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 0.10 \pm \\ 0.01 \end{array}$
P (mg/kg)	$\begin{array}{c} 7.02 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 67.07 \pm \\ 3.20 \end{array}$	$\begin{array}{r} 76.45 \pm \\ 5.90 \end{array}$	$\begin{array}{r} 41.29 \pm \\ 3.80 \end{array}$	75.65 ± 6.10	$\begin{array}{c} 64.26 \pm \\ 7.00 \end{array}$	$57.33 \pm \\ 6.40$	$\begin{array}{c} 13.75 \pm \\ 1.80 \end{array}$	$\begin{array}{c} 20.09 \pm \\ 8.10 \end{array}$	101.55 ± 10.00	28.9 ± 7.40
Ca (cmol/kg)	$\begin{array}{c} 7.20 \pm \\ 0.80 \end{array}$	$\begin{array}{c} 6.90 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 5.70 \pm \\ 1.20 \end{array}$	$\begin{array}{c} 5.85 \pm \\ 0.90 \end{array}$	6.20 ± 1.90	$\begin{array}{c} 4.80 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 6.55 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 6.80 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 5.90 \pm \\ 1.30 \end{array}$	$\begin{array}{c} 6.34 \pm \\ 2.00 \end{array}$	6.40 ± 1.40
Mg (cmol/kg)	$\begin{array}{c} 4.55 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 4.83 \pm \\ 0.40 \end{array}$	4.25 ± 1.00	$\begin{array}{c} 4.70 \pm \\ 0.60 \end{array}$	$\begin{array}{c} 3.92 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 4.34 \pm \\ 0.90 \end{array}$	$\begin{array}{c} 4.50 \pm \\ 1.00 \end{array}$	$\begin{array}{c} 3.93 \pm \\ 0.80 \end{array}$	$\begin{array}{c} 4.30 \pm \\ 0.80 \end{array}$	$\begin{array}{c} 5.52 \pm \\ 0.40 \end{array}$	$\begin{array}{c} 5.32 \pm \\ 0.60 \end{array}$
Na (cmol/kg)	$\begin{array}{c} 0.34 \pm \\ 0.50 \end{array}$	$\begin{array}{c} 0.33 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 0.27 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.48 \pm \\ 0.20 \end{array}$	$\begin{array}{c} 0.30 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 0.27 \pm \\ 0.00 \end{array}$	0.41 ± 0.12	$\begin{array}{c} 0.37 \pm \\ 0.19 \end{array}$	$\begin{array}{c} 0.24 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.34 \pm \\ 0.10 \end{array}$	$\begin{array}{c} 0.30 \pm \\ 0.10 \end{array}$
K (cmol/kg)	$\begin{array}{c} 0.18 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.14 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.18 \pm \\ 0.07 \end{array}$	$\begin{array}{c} 0.17 \pm \\ 0.09 \end{array}$	$\begin{array}{c} 0.20 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.17 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.12 \pm \\ 0.03 \end{array}$	$\begin{array}{c} 0.20 \pm \\ 0.09 \end{array}$	$\begin{array}{c} 0.16 \pm \\ 0.09 \end{array}$	$\begin{array}{c} 0.20 \pm \\ 0.08 \end{array}$	$\begin{array}{c} 0.18 \pm \\ 0.00 \end{array}$
CEC (cmol/kg)	12.27 ± 0.52	$\begin{array}{c} 12.20 \pm \\ 1.00 \end{array}$	10.40 ± 1.40	11.20 ± 1.90	10.62 ± 1.50	$\begin{array}{c} 9.58 \pm \\ 2.00 \end{array}$	$\begin{array}{c} 11.58 \pm \\ 3.00 \end{array}$	$\begin{array}{c} 11.30 \pm \\ 0.00 \end{array}$	10.60 ± 1.70	12.40 ± 1.40	12.27± 0.00

Discussion

The soils were moderately acidic, with pH range of 5.09 - 6.39. The moderately acidic characteristic is typical of the Niger Delta soils (Odu et al., 1985; Isirimah, 1987). The soils were sandy loam, with sand fractions ranged between 80.10 - 93.10%, silt fraction; 4.50 - 10.00% and clay fractions range were from 1.40 - 9.90%. The cation exchange capacities (CEC) ranged from 9.58 cmol/kg in sample F to 12.40 cmol/kg recorded for sample J. The control value was 12.27 cmol/kg. The values obtained for CEC were largely due to Ca and Mg. The values for Ca accounted for more than half, while Mg accounted for over one third of the total CEC for each site including the control. The total organic carbon contents ranged between 0.07% (Sample I) and 1.28% (Sample F). The values recorded for the controls was $0.06 \pm 0.1\%$ The slight differences in CEC values in the soil may be explain in terms of slight differences in organic matter and clay content. Many authors (Sposito, 1989; Sparks, 2003; Rodriguez - Rubio et al., 2003; Yin et al., 2012) have correlated organic matter content to soil CEC. The low clay content and organic matter tend to make the soils permeable. The phosphorus content of the soils ranged between 7.02 - 101.55 mg/kg against a control value of 28.94 mg/kg.

The concentrations of heavy metals obtained from this study generally decreased vertically with soil depth. This higher level of metals on the top or surface soil is expected since the top soil is the point of contact. The surface soils therefore represent better indicators of metallic burdens since the metal levels decreased with depth in all the sites. The average top - soil value for Zn (764.55 mg/Kg) obtained in this study was higher than the critical soil total content of Zn (Table 5) for the several countries indicating pollution due to automobile artisans' activities. The value was however lower than the 1086 mg/Kg Zn recorded by Nwoko *et al* (2012) for auto - mechanic villages in Imo State, Southeast Nigeria. The range (400.01 - 1126.04 mg/Kg) for the top - soil (Table 6) is comparable to that obtained in Akure (391.57 - 782.23 mg/Kg) (Oguntimehin and Ipinmoroti, 2008). High

concentration of zinc kills or stunts plant growth. In human, excessive intake of Zn may aggravate Cu deficiency (Hurley and Keen, 1987). The abundance of Zn in the soil cannot be ascribed to its relative natural occurrence in soil; anthropogenic addition

from the metal junkyard found in many of the auto mechanic sites is probable. Used oils that sink into the ground as leachates also contain a lot of this metal.

Samples	Depths (cm)	Zn	Mn	Pb	Cu	Cr	Cd
A	0 - 15 15 - 30 30 - 45 45 - 60	615.57 ± 3.50 519.75 ± 5.10 445.24 ± 11.40 199.09 ± 10.20	$\begin{array}{c} 262.22 \pm 10.00 \\ 212.00 \pm 11.00 \\ 181.14 \pm 0.00 \\ 113.41 \pm 8.00 \end{array}$	$\begin{array}{c} 36.33 \pm 2.00 \\ 26,22 \pm 1.14 \\ 23.71 \pm 1.40 \\ 18.60 \pm 2.90 \end{array}$	$59.30 \pm 6.60 \\ 55.65 \pm 3.50 \\ 50.19 \pm 7.17 \\ 39.80 \pm 13.70$	$\begin{array}{c} 80.10 \pm 3.00 \\ 61.80 \pm 10.00 \\ 31.1010 \pm 4.00 \\ 11.60 \pm 1.00 \end{array}$	$\begin{array}{c} 9.30 \pm 0.00 \\ 6.78 \pm 0.00 \\ 3.05 \pm 0.00 \\ 0.65 \pm 0.00 \end{array}$
В	0 - 15 15 - 30 30 - 45 45 - 60	$\begin{array}{c} 400.01 \pm 9.80 \\ 392.00 \pm 12.50 \\ 336.00 \pm 20.00 \\ 277.43 \pm 14.00 \end{array}$	$\begin{array}{c} 159.00 \pm 12.00 \\ 73.38 \pm 5.00 \\ 66.08 \pm 9.00 \\ 34.09 \pm 4.90 \end{array}$	$\begin{array}{c} 37.40 \pm 2.90 \\ 32.36 \pm 2.00 \\ 20.00 \pm 3.00 \\ 11.13 \pm 00 \end{array}$	$50.04 \pm 10.00 47.30 \pm 12.00 38.40 \pm 14.00 30.10 \pm 14.00$	$\begin{array}{c} 87.10 \pm 14.00 \\ 61.31 \pm 3.00 \\ 36.10 \pm 2.00 \\ 11.60 \pm 1.00 \end{array}$	$\begin{array}{c} 9.35 \pm 1.00 \\ 5.90 \pm 1.00 \\ 3.40 \pm 0.50 \\ 1.40 \pm 0.00 \end{array}$
С	0 - 15 15 - 30 30 - 45 45 - 60	$744.32 \pm 7.17 599.88 \pm 12.10 474.41 \pm 13.10 203.48 \pm 0.00$	$\begin{array}{c} 179.45 \pm 23.00 \\ 137.79 \pm 12.10 \\ 107.78 \pm 10.00 \\ 79.00 \pm 2.80 \end{array}$	$\begin{array}{c} 35.93 \pm 0.00 \\ 37.21 \pm 0.00 \\ 33.58 \pm 2.00 \\ 16.93 \pm 1.70 \end{array}$	$\begin{array}{l} 89.86 \pm 22.00 \\ 93.04 \pm 20.40 \\ 85.88 \pm 30.00 \\ 55.98 \pm 12.00 \end{array}$	$77.31 \pm 12.00 \\ 60.88 \pm 14.40 \\ 37.19 \pm 0.00 \\ 18.60 \pm 2.00$	$\begin{array}{c} 8.95 \pm 1.55 \\ 8.40 \pm 1.00 \\ 5.65 \pm 1.60 \\ 4.05 \pm 0.50 \end{array}$
D	0 - 15 15 - 30 30 - 45 45 - 60	$\begin{array}{l} 1016.25 \pm 24.50 \\ 842.68 \pm 10.20 \\ 656.20 \pm 14.50 \\ 522.67 \pm 17.14 \end{array}$	$131.08 \pm 4.70 \\ 108.12 \pm 5.00 \\ 48.12 \pm 3.00 \\ 34.48 \pm 3.80$	$50.82 \pm 5.00 \\ 35.20 \pm 4.40 \\ 21.34 \pm 1.20 \\ 10.51 \pm 8.20$	$\begin{array}{c} 155.03 \pm 20.00 \\ 127.33 \pm 18.10 \\ 130.35 \pm 40.00 \\ 112.49 \pm 20.00 \end{array}$	55.15 ± 5.00 31.57 ± 3.00 20.65 ± 3.00 10.30 ± 3.00	$\begin{array}{c} 6.80 \pm 1.00 \\ 3.77 \pm 0.05 \\ 1.55 \pm 0.50 \\ 0.83 \pm 0.00 \end{array}$
Ε	0 - 15 15 - 30 30 - 4 45 - 60	$\begin{array}{c} 1126.04 \pm 10.00 \\ 783.52 \pm 8.90 \\ 618.08 \pm 9.40 \\ 520.00 \pm 1.00 \end{array}$	$\begin{array}{c} 191.79 \pm 14.40 \\ 148.42 \pm 7.50 \\ 103.05 \pm 4.50 \\ 48.22 \pm 2.30 \end{array}$	$\begin{array}{c} 48.22 \pm 10.00 \\ 39.15 \pm 3.40 \\ 28.34 \pm 6.00 \\ 11.75 \pm 3.00 \end{array}$	$\begin{array}{c} 120.58 \pm 11.00 \\ 97.94 \pm 12.40 \\ 91.20 \pm 8.00 \\ 80.50 \pm 11.00 \end{array}$	$\begin{array}{c} 24.11 \pm 2.10 \\ 19.59 \pm 1.00 \\ 9.72 \pm 0.80 \\ 5.19 \pm 0.00 \end{array}$	$\begin{array}{c} 8.60 \pm 1.50 \\ 6.09 \pm 1.50 \\ 4.29 \pm 0.50 \\ 0.97 \pm 0.00 \end{array}$
F	$\begin{array}{r} 0 & - & 15 \\ 15 & - & 30 \\ 30 & - & 45 \\ 45 & - & 60 \end{array}$	$\begin{array}{c} 796.36 \pm 5.50 \\ 672.44 \pm 10.20 \\ 605.56 \pm 7.50 \\ 396.16 \pm 11.00 \end{array}$	$181.98 \pm 0.00 \\ 138.72 \pm 0.00 \\ 99.97 \pm 10.20 \\ 58.29 \pm 7.70$	39.32 ± 2.00 39.80 ± 2.00 27.86 ± 4.50 16.36 ± 2.00	$\begin{array}{l} 99.54 \pm 20.00 \\ 84.68 \pm 12.00 \\ 54.59 \pm 10.00 \\ 46.88 \pm 10.00 \end{array}$	$\begin{array}{c} 20.20 \pm 2.00 \\ 36.90 \pm 4.00 \\ 13.69 \pm 2.20 \\ 9.90 \pm 1.10 \end{array}$	$\begin{array}{c} 8.18 \pm 2.25 \\ 5.45 \pm 1.50 \\ 3.49 \pm 0.50 \\ 1.95 \pm 0.50 \end{array}$
G	$\begin{array}{r} 0 & - & 15 \\ 15 & - & 30 \\ 30 & - & 45 \\ 45 & - & 60 \end{array}$	$780.00 \pm 11.30 \\ 506.81 \pm 20.00 \\ 396.04 \pm 12.80 \\ 218.03 \pm 10.30$	$\begin{array}{c} 176.20 \pm 9.80 \\ 173.52 \pm 10.20 \\ 152.40 \pm 10.00 \\ 102.30 \pm 9.80 \end{array}$	$\begin{array}{c} 35.34 \pm 4.30 \\ 24.40 \pm 10.00 \\ 15.24 \pm 8.50 \\ 7.20 \pm 3.70 \end{array}$	$\begin{array}{l} 88.10 \pm 0.00 \\ 77.55 \pm 0.00 \\ 55.35 \pm 5.00 \\ 47.99 \pm 4.40 \end{array}$	$\begin{array}{c} 62.70 \pm 5.70 \\ 51.55 \pm 12.00 \\ 31.70 \pm 10.00 \\ 20.16 \pm 5.40 \end{array}$	$\begin{array}{c} 8.80 \pm 1.00 \\ 6.08 \pm 1.50 \\ 3.81 \pm 0.55 \\ 1.35 \pm 0.00 \end{array}$
Н	$\begin{array}{r} 0 & - & 15 \\ 15 & - & 30 \\ 30 & - & 45 \\ 45 & - & 60 \end{array}$	$\begin{array}{c} 615.04 \pm 15.40 \\ 499.96 \pm 22.00 \\ 432.28 \pm 17.30 \\ 318.92 \pm 10.00 \end{array}$	$\begin{array}{c} 179.05 \pm 14.20 \\ 188.24 \pm 17.30 \\ 172.23 \pm 13.10 \\ 138.72 \pm 12.00 \end{array}$	$\begin{array}{c} 30.78 \pm 1.00 \\ 25.99 \pm 3.00 \\ 19.87 \pm 0.00 \\ 9.89 \pm 0.00 \end{array}$	$76.94 \pm 10.00 \\ 64.94 \pm 11.00 \\ 54.69 \pm 14.14 \\ 46.97 \pm 17.00$	38.95 ± 4.80 29.94 ± 3.80 18.55 ± 2.00 9.21 ± 1.40	$\begin{array}{c} 9.97 \pm 1.55 \\ 6.22 \pm 1.55 \\ 4.76 \pm 1.05 \\ 2.11 \pm 0.50 \end{array}$
Ι	$\begin{array}{r} 0 & - & 15 \\ 15 & - & 30 \\ 30 & - & 45 \\ 45 & - & 60 \end{array}$	$\begin{array}{c} 755.27 \pm 8.90 \\ 575.12 \pm 14.70 \\ 473.73 \pm 18.20 \\ 364.64 \pm 12.30 \end{array}$	$\begin{array}{c} 165.76 \pm 12.10 \\ 139.82 \pm 10.00 \\ 86.34 \pm 9.00 \\ 57.76 \pm 8.40 \end{array}$	37.76 ± 0.90 27.56 ± 10.00 18.20 ± 5.00 12.40 ± 2.00	$\begin{array}{l} 94.41 \pm 10.00 \\ 89.17 \pm 13.70 \\ 71.41 \pm 10.00 \\ 44.19 \pm 5.00 \end{array}$	$\begin{array}{c} 34.17 \pm 1.20 \\ 21.88 \pm 1.00 \\ 12.34 \pm 2.00 \\ 8.12 \pm 1.10 \end{array}$	$\begin{array}{l} 9.41 \pm 1.00 \\ 6.69 \pm 0.75 \\ 4.14 \pm 0.60 \\ 1.55 \pm 0.50 \end{array}$
J	$\begin{array}{r} 0 & - & 15 \\ 15 & - & 30 \\ 30 & - & 45 \\ 45 & - & 60 \end{array}$	$796.64 \pm 30.00 \\ 643.60 \pm 32.40 \\ 467.96 \pm 21.50 \\ 383.52 \pm 14.00$	$\begin{array}{c} 192.37 \pm 12.80 \\ 121.89 \pm 10.00 \\ 93.80 \pm 10.20 \\ 63.34 \pm 4.80 \end{array}$	39.85 ± 3.90 32.80 ± 2.80 19.60 ± 3.10 10.95 ± 2.00	$\begin{array}{c} 99.58 \pm 0.00 \\ 95.23 \pm 12.00 \\ 58.52 \pm 11.00 \\ 23.99 \pm 3.30 \end{array}$	$23.90 \pm 4.90 \\ 19.46 \pm 3.40 \\ 11.60 \pm 3.00 \\ 8.40 \pm 2.10$	9.58 ± 1.50 7.30 ± 1.55 4.70 ± 1.00 2.13 ± 0.50
x	$\begin{array}{r} 0 & - & 15 \\ 15 & - & 30 \\ 30 & - & 45 \\ 45 & - & 60 \end{array}$	$\begin{array}{c} 11.80 \pm 1.40 \\ 6.60 \pm 0.90 \\ 4.20 \pm 1.00 \\ 1.00 \pm 0.40 \end{array}$	$\begin{array}{c} 4.49 \pm 1.10 \\ 2.80 \pm 0.90 \\ 1.984 \pm 0.30 \\ 1.244 \pm 0.30 \end{array}$	$\begin{array}{c} 1.45 \pm 0.00 \\ 0.75 \pm 0.10 \\ 0.44 \pm 0.00 \\ 0.15 \pm 0.00 \end{array}$	$\begin{array}{c} 15.58 \pm 3.00 \\ 7.55 \pm 0.00 \\ 1.47 \pm 0.00 \\ 0.57 \pm 0.00 \end{array}$	$\begin{array}{c} 0.20 \pm 0.00 \\ 0.14 \pm 0.00 \\ 0.06 \pm 0.10 \\ < 0.04 \end{array}$	$\begin{array}{c} 0.02 \pm 0.00 \\ < 0.01 \\ < 0.01 \\ < 0.01 \end{array}$

The average top - soil value of 181.89 mg/Kg Mn obtained in this study was lower than the concentration recorded in Yauri (608.11 mg/Kg) for road side soils (Yahaya *et al.*, 2009). The value compared with similar studies from other countries is lower than that recorded in the United States (2532 mg/Kg), China (1740 mg/Kg), and Poland (1122 mg/Kg) (Dudka, 1992; Bradford *et al.*, 1996; Abida *et al.*, 2009). Manganese though an essential element, may be toxic at elevated concentrations. The toxic effect in humans is associated with severe psychiatric disorder resembling

schizophrenia, followed by permanently crippling neurological disorder clinically similar to Parkinson's disease (Klaassen, 2001).

Lead in the soils has an average top - soil value of 39.18 mg/Kg. This value was higher than the critical soil Pb content for Canada (25 mg/Kg), Finland (38 mg/Kg), and Eastern Europe (32 mg/Kg); but lower than those of Denmark (40 mg/Kg), Czech Republic (70 mg/Kg), Netherland (85 mg/Kg), and Switzerland and Ireland (50 mg/Kg). The value was also lower than those obtained in Kaduna (76.92 mg/Kg), Iwo (126 mg/Kg), Makurdi (123 mg/Kg), and Owerri (1162 mg/Kg) (Okunola *et al.*, 2007; Ayodele and Dawodu, 2008; Nwachukwu *et al.*,2011; Pam *et al.*,2013). The upper limit of the range (30.78 - 50.82 mg/Kg) obtained here are much lower those of 70.80 - 192.03 mg/Kg, 11.75 - 2882.50 mg/Kg and 99.00 - 1090.00 mg/Kg Pb ranges recorded by Oguntimehin and Ipinmoroti (2008), Adelekan and Abegunde

(2011), and Nwachukwu et al (2010) for Akure, Ibadan, and Okigwe respectively. Lead has no known physiological relevant role in body and its harmful effects are wide ranging (White *et al.*, 2007). Lead impaired mental and physical development, decrease heme biosynthesis, elevate hearing threshold and decrease serum levels of vitamin D (Needleman, 1990).

Table 4: The average metal content of soil samples at the different depths in Sapele automobile workshops

			Result ± SD			
Depths (cm)	Zn (mg/kg)	Mn (mg/kg)	Pb (mg/kg)	Cu (mg/kg)	Cr (mg/kg)	Cd (mg/kg)
0 - 15	764.55 ± 12.60	181.89 ± 11.30	39.18 ± 3.20	93.34 ± 10.96	50.40 ± 5.47	8.89 ± 1.24
15 - 30	603.58 ± 14.70	144.19 ± 9.09	32.07 ± 3.87	83.28 ± 12.01	39.49 ± 5.56	6.26 ± 1.04
30 - 45	490.55 ± 16.39	111.09 ± 7.90	22.77 ± 3.47	69.06 ± 14.93	22.26 ± 2.90	3.88 ± 0.67
45 - 60	340.40 ± 10.02	72.96 ± 6.45	12.57 ± 2.55	52.89 ± 11.04	11.31 ± 1.81	1.70 ± 0.25

The average top - soil value for Cu was higher than the critical soil total Cu content for the several countries. The value was also higher than the 24.60 mg/Kg Cu obtained for North Bank mechanic village in Makurdi by Pam et al (2013). It was however lower than the 1004 mg/KgCu recorded by Nwoko et al (2012) for soil in auto - mechanic village in Southeast region of Nigeria. The range (50.04 - 155.03 mg/kg Cu) obtained in this study is within the 40.72 - 151.43 mg/Kg and 1.48 - 476.00 mg/Kg obtained by Oguntimehin and Ipinmoroti (2008); Adelekan and Abegunde (2011), but was lower than the 102.00 - 1001.00 mg/Kg Cu recorded by Nwachukwu et al (2010). Copper, though an essential element may be harmful at high concentration. Copper effects on plants physiology are wide ranging, including interference with fatty acid and protein metabolism and inhibition of respiration and nitrogen fixation process. At the whole plant levels Cu is an effective inhibitor of vegetative growth and induces general symptom of senescence (Fernandez and Henrigues, 1991; Konstantinidis et al., 2003). In mammals excessive dietary copper produces nausea, vomiting and diarrhoea as well as pathological changes in brain tissue (Pizzarro et al., 1999). A spongy transformation of the white matter of the Midbrain and Cerebellum in lambs has long been reported (Doherty et al, 1969). Copper accumulates in livers, kidney, cornea, and brain. The accumulation in brain causes trauma, which lead to death (Davis et al., 2000; Varela - Nallar et al., 2006).

Chromium average top - soil value was 50.40 mg/Kg and was only higher than the critical soil total Cr content for Canada (20 mg/Kg) and Denmark (50 mg/Kg). The upper (87.10 mg/Kg) and least (20.20 mg/Kg) limits obtained in this study were both higher than the 20 mg/Kg critical soil total Cr content recommended in Canada. The range in this study was also higher than the 4.00 - 27.00 mg/Kg Cr obtained by Nwachukwu *et al* (2010) within and around mechanic villages in South - east Nigeria. Chromium though an essential element may be toxic at high levels. Chromium (VI) has been reported to exhibit mutagenic, carcinogenic and teratogenic effects (Cheng and Dixon, 1998; Asmatullah *et al*, 1998; Kaltreider *et al*, 1999; Soils - Heredia *et al.*, 2000).

The average Cd content for the top - soils was 8.89 mg/Kg. This value was higher than the critical soil total Cd for several countries (Table 5). This is an indication of pollution by cadmium in the soil. The value was much lower than the 43.00 mg/Kg Cd obtained by Nwoko et al (2012), but much higher than the 0.60 mg/Kg recorded by Pam et al (2013) for auto - mechanic villages in Makurdi. The range of value obtained (6.80 - 9.41 mg/Kg) is within the range (0.43 - 17.23 mg/kg) recorded by Adelekan and Abegunde (2011) in Ibadan. Shear off from metal plating, leachates from used oils and old tyres frequently burnt on these sites are likely source of cadmium (Kabata - Pendias, 1995; Rabie et al., 1996). Cadmium has no constructive purpose in human body. Cadmium and its compounds are extremely toxic even in low concentration, and will bio - accumulate in organisms and ecosystem. Ingestion of any significant amount of Cadmium causes immediate poisoning and damage to the liver and the kidneys. Compounds containing cadmium are also carcinogenic (Kalonel, 1976; Kazantzis, 1979; Fasanya - Odewumi et al., 1998; II'yasova, 2005; Sahmoun et al., 2005). Ingestion of Cadmium in excess leads to a disease known as Itai - Itai, a disease resulting in soft bones, shrinking body and death (Ademoroti, 1996; Kazantzis, 2004).

Table 5: Critical soil tota	contents of heav	y metals for several	countries
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Countries		Critical limit (mg/kg)									
	Zn	Pb	Cu	Cr	Cd						
Canada	50	25	30	20	0.5						
Denmark	100	40	30	50	0.3						
Finland	90	38	32	80	0.3						
Czech republic	150	70	70	130	0.4						
Netherland	140	85	36	100	0.8						
Switzerland	200	50	50	75	0.8						
Ireland	150	50	50	100	1.0						
Eastern Europe	100	32	55	90	0.2						

(Eastern Europe include Russia, Ukraine, Moldova and Belarus)

Source: De Vries and Bakker, 1998

To explain the heavy metal profile on the locations, an appraising method by calculating the contamination/ pollution (C/P) index values for each metal in each location was calculated using the Canadian critical values for heavy metals in soil (Table 5). This index represents the metal content effectively measured in soil by chemical analysis and the critical values from a reference table (Lacatusu, 1998). Table 7 show the C/P indices distribution for different depths in all locations.

Unlike the pattern observed in Table 3, there is much difference between the concentration of metal in soil and the potential treat pose by them expressed though their c/p index value. Lacatusu (1998) established that the c/p value index is directly proportional to their level of contamination and pollution of specific metal species in each case. Higher value > 1 indicate higher risk while lower values < 1 connote minimal risk to the environment (Table 8).

Manganese (Mn) is not in the lists of standards critical soil total contents of heavy metals for several countries. Manganese, though an essential metals is toxic to a varying degree, depending upon the type of manganese ion and its oxidation state. Permanganates appear to be a highly toxic form of manganese because of their high solubility and powerful oxidizing action on the tissues (Bouchard et al., 2007).

Tal	ble 6	Rai	nge of	f tota	l meta	l cor	ntent (mg	/kg) of to	p-soil	l sam	ples i	n Sa	pele	autom	obile	work	shop	s relat	ive t	o othe	r areas
								(<i>0</i>		,	P												

Metal	Values of stud	y area (Sapele)	Akure ^a	Ibadan ^b		
Zn	Min.	400.01	391.57			
	Max.	1126.04	782.23			
Mn	Min.	131.08				
	Max.	262.22				
Pb	Min.	30.78	70.80	11.75		
	Max.	50.82	192.03	2882.50		
Cu	Min.	50.04	40.72	1.48		
	Max.	155.03	151.43	476.00		
Cr	Min.	20.20	21.50	5.50		
	Max.	87.10	181.80	29.75		
Cd	Min	6.80		0.43		
	Max.	9.41		17.23		

Source: a = Oguntimehin and Ipinmoroti (2008), b = Adelekan and Adegunde (2011)

Samulas		Results (Depths, cm)									
Samples	0 – 15	15 - 30	30 - 45	45 - 60	Total						
Α	38.35	29.95	19.18	7.93	95.41						
В	34.22	23.58	17.41	10.38	85.59						
С	41.08	36.43	26.85	15.64	120.00						
D	43.88	31.69	22.46	16.80	114.83						
Е	46.87	33.66	25.60	15.75	121.88						
F	38.19	30.61	22.71	14.54	106.05						
G	40.69	28.43	19.58	9.96	98.66						
Н	37.98	27.14	21.71	13.02	99.85						
Ι	40.29	30.05	21.48	12.77	104.59						
J	41.20	32.93	22.07	13.59	109.79						
Х	0.86	0.42	0.15	0.05	1.48						

Table 7: Heavy metal pollution profile in Sapele, calculated from total meta	l C/P index	
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Table 7 shows the profile of the heavy metals in the soils. To obtain a profile the total pollution ranges defined by multiple pollution found from contribution from each metal contamination/ pollution status were used. The observed trend for all samples shows that the top - soils are heavily loaded with heavy metals than the sub - soils. Location E had the highest total pollution by heavy metals with value of 121.88. The values for the other locations are A (95.41), B (85.59), C (120.00), D (114.83), F (106.05), G (98.66), H (99.85), I (104.59), and J (109.79). These values are much higher than the control value (1.48). The degree of metal pollution in the locations was in the order: Oleh Road > Reclamation/Green - Egbede Road > Awolowo/Okpe Road > Okerigwre Road > Akintola Road > Shell Road > Gana New Road > Ajogodo Road > Ogodo Road > McPherson Road Characterization based on the significance of intervals of

contamination/pollution (C/P) index in Table 8, reveals that due to multiple pollutions, all the locations are severely polluted with respect to the five heavy metals.

Figure 2 shows the graph of c/p index versus depth for each metal calculated from the average concentrations of the metals using the Canadian critical values for heavy metals in soil. The soils were polluted at all depths with zinc, cadmium and copper. Depths 0 - 15 and 15 - 30 cm too were polluted with Cr; while only the topmost soils (0 - 15 cm) were polluted with lead. Generally, the metals contents decreases with increase in depths. The relative high concentrations found for most metals even at depth 45 - 60 cm signifies that heavy metals in these locations are not restricted to top - soil (0 - 15 cm), and that there are possibilities of leaching of these heavy metals to underground water and subsequent effects on public health.

C/P	Significance
<0.1	Very slight contamination
0.10-0.25	Slight contamination
0.26-0.50	Moderate contamination
0.51-0.75	Severe contamination
0.76-1.0	Very severe contamination
1.1-2.0	Slight pollution
2.1-4.0	Moderate pollution
4.1-8.0	Severe pollution
8.1-16.0	Very severe pollution
>16	Excessive pollution

 Table 8: Significance of intervals of contamination/pollution (C/P) index

Source: Lacatusu, 1998



Figure 2: Plot of C/P versus depth for heavy metals

Statistical analysis

The analytical results were compiled to form a multi - element data base using excel as statistical software. Pearson's correlation coefficients and P - values were calculated for all possible variable pairs (Tables 9 - 10). The total contents of metals were negatively correlated with soil depth. Very significant positive correlations were present between Zn/Pb, Zn/Cu, Mn/Cd, Pb/Cd, and Cr/Cd. Significant positive correlation were also observed for Zn/Cd, Mn/Pb, Mn/Cr, Pb/Cu, and Pb/Cr in soils samples. These positive correlations among metals tend to suggest same source. Insignificant positive correlations were found between Zn/Mn, Mn/Cu, Zn/Cr, Cu/Cr, and Cu/Cd. These insignificant positive correlations between metals indicate that the appearance of local high concentration for metal by possible contamination does not necessarily indicate high values for others metals. Slight correlations (except for few instances) between soil properties (clay, organic matter content, cation exchange capacity etc.) and the total concentration of metals were found.

	Zn	Mn	Pb	Cu	Cr	Cd			
Zn	1.000	0.427	0.8147**	0.8374**	0.2419	0.5974*			
Mn		1.000	0.5862*	0.1549	0.5674*	0.7605**			
Pb			1.000	0.6264*	0.5626*	0.8144**			
Cu				1.000	0.1161	0.3582			
Cr					1.000	0.6729**			
Cd						1.000			
**significant at the %1 level *significant at the %5 level									

**significant at the %1 level, *significant at the %5 level

siguine matter and eation exemange capacity (independent variables) in son samples.									
	Zn	Mn	Pb	Cu	Cr	Cd			
Soil depths (cm)	-0.7063**	-0.7249**	-0.8602**	-0.4956	-0.6803**	-0.9237**			
Clay (%)	-0.7627**	0.4539	-0.5166	-0.4278	0.6996**	0.4979			
Silt (%)	-0.6314*	0.1408	-0.1770	-0.4369	0.4391	0.5984			
Sand (%)	0.7875**	-0.3541	0.4111	0.4820	-0.6540*	-0.6050*			
pH	0.0260	0.0035	-0.4387	-0.1377	-0.6173	0.0834			
Organic matter (%)	0.3969	-0.0003	0.5655	0.1523	-0.0396	0.0124			
CEC	-0.4633	0.1794	-0.5522	-0.3423	0.3366	-0.0145			

Table 10: Coefficient of relations between heavy metal content (dependent variable) and soil depth, clay, silt, sand, organic matter and cation exchange capacity (independent variables) in soil samples.

**significant at the %1 level, *significant at the %5 level

Conclusion

Type of contaminant, concentration and vertical distribution in soil are required inputs for development of predictive models, effective remediation and risk assessment. The concentration of heavy metals (Zn, Mn, Pb, Cu, Cr and Cd) found for soil in the vicinity of auto mechanic workshops were much higher than that from the control source. This is an indication that mechanic workshops situated within the metropolis are anthropogenic sources of heavy metals in soil. The concentrations of Zn, Cu, and Cd at most sites and depths were also higher than the recommended criteria values for several countries. Lead and Chromium have their top - soil (0 - 30 cm) values higher than most established criteria values but the sub - soil (30 - 60 cm) were within the stipulated values. The relative high concentrations found for most metals even at depth 45 - 60 cm, signifies that heavy metals in these sites are not restricted to top - soil (0 - 15 cm), and that there are possibilities of leaching of these heavy metals to underground water and subsequent effects on public health. Restriction should be imposed on auto - mobile workshops artisans to prevent them from dumping their waste indiscriminately and proper enlightenment carried out, to alert all stake - holders on the dangers inherent in indiscriminate dumping. Auto mobile workshops soil should be heavily cemented and proper drainage constructed to channel used oils, lubricants and spilled gasoline to a reservoirs where they can be properly treated or chemically re - processed to harness these heavy metals in them.

References

Abida, B., Ramaih, M., Harishma, I.K., and Veena, K. (2009). Analysis of heavy metals concentrations in soils and litchens from various localities of Hosur road, Bangalore, India, E. Journals of Chemistry, 1: 13 - 22

Adelekan, B.A and Abegunde, K.D. (2011). Heavy metals contamination of soil and groundwater at automobile mechanic villages in Ibadan, Nigeria. Int. J. Phys. Sci., 6(5): 1045 - 1058.

Ademoroti, C.M.A. (1996). Environmental chemistry and toxicology, Foludex press, Ibadan. Pp 192 – 199

Aller, N. (1989). Handbook of Suggested Practices for Design and Installation of Groundwater, Monitoring Wells. NWWA. Dubling. EPA/600/4 - 89/034.

Alloway, B. J. and Ayres, D.C. (1997). Chemical Principles of Environmental pollution. Blackie Academic and professional, Glasgow. Pp 190 - 217.

Asmatullah, Qureshi, S.N. and Shakoori, A.R. (1998). Hexavalent Chromium induced congenital abnormalities in Chick embryos. J. Appl. Toxicol., 18(3): 167–171.

Ayodele, R.I. and Dawodu, M. (2008). Heavy metals contamination of topsoil and dispersion in the vicinities of reclaimed auto - repair workshops in Iwo, Nigeria Bull. Chem. Soc. Ethiop., 22(3): 339 - 348.

Biney, C. A; Amuzu, A.T; Calamarr, D; Kaba, I. L; Haeve, H; Ochumba, P. B. O; Osibanjo, O; Radegonde V. and Saad, M. A.

H. (1994). Review of heavy metals, In: Calamari, D. and Naeve, H. (eds.) Review of pollution in African Aquatic Environment FOA/CIFA Technical paper 25, pp. 30 - 60.

Bouchard, M., Laforest, F., Vandelac, L., Bellinger, D and Mergler, D. (2007) Hair manganese and hyperactive behaviours, Pilot study of school - age children exposed through tap water. Environ. Health Perspert. , 115(1): 122 - 127.

Boulding, J. R. (1994). Description and sampling of contaminated soils. 2nd edn, Lewis publishers, New York, No. A10, pp.11 - 23

Bouyoucos, G.J. (1962). Improved hydrometer method for making particle size analysis of soils. Agron. J. 54:464 - 465.

Bradford, G.R., Chang, A.C., and Page, A.L. (1996). Background concentrations of trace and major elements in California soils, Kearney Foundation Special Report, University of California, Riverside, Pp. 1 - 52

Cheng, L and Dixon, K. (1998). Analysis of rapair and mutagenesis of chromium induced DNA damage in yeast mammalian cells and transgenic mice. Environ. Health Perspect. 106:1027 – 1032.

Davis, J.A., Volesky, B., Vierra, R.H. (2000). Sargassum seaweed as biosorbent for heavy metals. Water Res. 34(17): 4270 - 4278.

De Vries and Bakker, (1998). Manual for calculating critical loads of heavy metals for terrestrial ecosystem: Guidelines for critical limits, calculation methods and input data TNO institute of Environmental Sciences, Energy Research and Process Innovation. Den Helder, The Netherlands. pp. 144

Doherty, P.C., Barlow, R.M. and Angus, K.W. (1969). Spongy changes in the brain of sheep poison by excess Dietary copper, Res. Vet. Sci. 10: 303 - 304.

Dudka, S. (1992). Factor analysis of total element concentrations in surface soils of Poland, Sci. Tatal Environ. 121: 39 - 52.

E. E. A. (2007). Progress in Magement of contaminated sites (CSI 015) Europe Environmental Assessment Agency. Published in July 2005; Kongan, 6DK - 1050, Denmark, http://www.reea.europa.eu

Fasanya - Odewumi, C., Latinwo, L.M., Ikediobi, C.O., Gillard, L., Sponholtz, G., Nwoga, J., Stino, F., Hamilton, N., Erdos, G.W. (1998). The genotoxicity and cytotoxicity of dermally - administered Cadmium: effects of dermal Cadmium administration, Int. J. Mol. Med., 1(6): 1001 – 1006.

Fernandes, J. C and Henrigues, F.S. (1991). Biochemical physiological and structural effects of excess Copper in plants. Journal the Botanical review. 57(3): 246 – 273.

Folson, B.L., Lee, C.R. and Bastas, D.J. (1981). Influence of disposal environment on availability and plant uptake of heavy metals in dredged material. U.S. Army, Washington. Tech. Rep. El. 81-12

Hurley, L.S. and Keen, C.L. (1987). Trace Elements in Human and Animal Nutrition, Mertz, W(Ed). Academic press, New York, P. 185

Igwe, J. C. and Abia, A. A. (2006). A Bio - separation Process for Removing Heavy metals from Waste Water using Biosorbents. Afr. J. Biotechnol. 5(12):1167 - 1179.

Il'yasova, D.S.G.G. (2005). Cadmium and renal cancer, Toxicol. Appl. Pharmacol., 207(2): 179 – 186.

Isirimah, N.O. (1987). An inventory of some chemical properties of selected surface soils of Rivers state of Nigeria: In proceeding of 15th annual conference of Soil Science Association of Nigeria, Kaduna, 217 - 233.

Kabata - Pendias, A. (1995). Agricultural problems related to trace metal contents of soil, In: Heavy metals (Problems and Solutions) Salomons, W., Forstner, U., and Mader, P (Eds), springer Verlag, Berin, Pp. 3 - 18.

Kalonel, L.N. (1976). Association of cadmium with renal cancer, 37(4): 1782 – 1787.

Kaltreider, R.C., Pesce, C.A., Ihnat, M.A., Lariviere, J.P. and Hamilton J.W. (1999). Differential effects of arsenic (III) and chromium (VI) on nuclear transcription factor binding, Mol. Carcinog., 25: 219 - 229.

Kazantzis, G. (1979). Renal tubular dysfunction and abnormalities of Calcium metabolism in Cadmium workers. Environ. Health Perspect., 28: 155 – 159.

Kazantzis, G. (2004). Cadmium osteoporosis and calcium metabolism. Biometals, 17(5): 493 – 498.

Klaassen, C.D. (2001). Heavy metals and heavy metal antagonist. In: Hardman et al (Eds). The Pharmaceutical basis of therapeutics, McGraw Hill, New York, pp 1851 – 1875

Konstantinidis, K.T., Isaacs, N., Fett, J., Simpson, S., Long, D.T. and Marsh, T.L. (2003). Microbial diversity and resistance to copper in metal contaminated Lake Sediment. Microbiol. Ecol., 45: 191 - 202

Lacatusu, R, (1998). Appraising Levels of Soil Contamination and Pollution with Heavy Metals In: Land information System for Planning the Sustainable use of Land Resources.

Heinike, H.J; Eckelman, W; Thomasson, A. J; Jones, R. J. A. and Buckley, B (Eds) European Soil Bureau, Research report NO.4, office for official publications of the European Communities Luxembourg, pp. 393 - 402.

Neal, B. G; Lawrence, E. B; Weandt J. L. (1990). Alkali Metal Partitioning in Pulvarized Coal Combustion, Sci, Tech; 74: 211 - 214.

Needleman, H. l. (1990). The future challenge of Lead toxicity, Environ. Health Perspect. 89: 85 – 90.

Nelson, D. W; and Sommers, L.E. (1982). Total carbon and matter. In page, A.Z. et al. (Ed) Methods of soil analysis. Part 2; 2nd ed. ASA, SSSA, Pp 539 - 579.

Nwachukwu, M.A., Feng, H., and Alinnor, J. (2010). Assessment of heavy metals in soil and their implications within and around mechanic villages, Int. J. Environ. Sci. Tech. 7(2): 347 - 358.

Nwachukwu, M.A., Feng, H., and Alinnor, J. (2011). Trace metal deposition in soil from auto - mechanic village to urban residential areas in Owerri, Nigeria. Procedia Environ. Sci. 4: 310 - 322.

Nwoko, C.O., Onoh, C.P. and Onoh, G.O. (2012). Remediation of trace metal contaminated Auto - mechanic soils with mineral supplemented organic Amendments, Universal J. Environ. Research Technol., 2(6): 489 - 499.

Odu, C.T.I., Nwoboshi, I.C., Esuruoso, O.F. and Ogunwale, O.J. A. (1985). Industry Operation Areas. Proceeding of the international seminar on the petroleum and the Nigerian Environment. Port Harcourt, 117 - 123.

Oguntimehen, I and Ipinmoti, K. (2008). Profile of Heavy metals from automobile workshops in Akure, Nigeria. J. Environ. Sci. and Technol., 1(1); 19 - 26.

Okunola, O.J., Uzairu, A., and Ndukwe, G. (2007). Levels of trace metals in soil and vegetation along major and minor roads in metropolitan city of Kaduna, Nigeria, Afri. J. Biotech. 6(14): 1703 - 1709.

Olsen, S.R. and Sommers, L.E. (1982) Phosphorus In: Page, A.L., Miller, R.H. and Keeny, D.R. (Eds.), Methods of Soil Analysis, Part 2, Chemical and Microbiological Properties. Madison, Wisconsin: pp. 403 - 430

Pam, A.A., Sha'Ato, R., and Offem, J.O. (2013). Contributions of automobile mechanic sites to heavy metals in soil: A case study of North Bank Mechanic village Makurdi, Benue State, Central Nigeria, J. Chem. Bio. Phy. Sci. 3(3): 2337 - 2347.

Pizzarro, F., Olivares, M., Uauy, R., Contreras, P., Rebelo, A. and Gidi, V. (1999). Acute gastro - intestinal effects of graded levels of copper in drinking water, Environ. Heath Perspect. 107: 117 - 121.

Rabie, F., Rashad, M.H., Khader, M.Y. and Hussein, W (1996). Contents of biogenic and non - biogenic heavy metals in El - Saff soil as related to different pollution sources. Egypt J. Soil Sci. 36(1 - 4): 165 - 177.

Reeuwijk, L.P. (Ed). 1995. Procedure for soil Analysis, International soil Reference and information, 5th ed. ISRIC, Wageningen, the Netherlands. Tech. paper 9. Pp. 3 - 17.

Rodriguez - Rubio, R; Basta, N.T; Casteel, S.W; Armstrong, F.P. and Ward, D.C. (2003). Chemical extraction methods to assess bio - available arsenic in soil and solid media. J. Environ. Qual., 32:876 – 884.

Sahmoun, A.E., Case, L.D. Jackson, S.A. and Schwartz, G.G. (2005). Cadmium and prostate cancer: A critical epidemiologic analysis, Cancer Invest. 23(3): 256 - 263.

Soils - Heredia, M.J., Quintailla - Vega, B., Sierra - Santoyo, A., Harnandez, J.M., Brambia, E., Cebrian, M.E. and Albores, A. (2000). Chromium increases pancreatic metallothionein in rat. Toxicology, 142: 111 - 117.

Sparks, D.L. (2003). Environmental soil chemistry 2nd ed., Academic Press, San Diego, CA. pp. 739 - 768

Sposito, G. (1989). The chemistry of soil. Oxford University Press, New York. pp. 201 - 238

Tessier, A; Campbell, P.G.C. and Bisson, M. (1979). Sequential extraction procedures for the speciation of particulate trace metals. Anal. Chem., 51(7): 884 – 851.

Varela - Nallar, L., Toledo, E.M., Larrondo, L.F., Cabral, A.L., Martins, V.R. and Inestrosa, N.C. (2006). Induction of cellular prion protein gene expressin by copper in neurons. Am. J. Physiol. Cell. Physiol., 290: 271 - 281.

White, D., Corg Slechta, A., Gilbert, E., Tiffany - Castighoni, E., Zawia, H., Virgolini. MyRoss - George, A., Laslem, M. (2007). New and evolving concept in Neuro - toxicology of Lead: Toxicology and Applied Pharmacology. 255(1): 1 - 27.

Yahaya, M. I., Ezeh, G. C., Musa, Y. F., and Mohammad, S. Y. (2009). Analysis of heavy metals concentration in road side soils in Yauri, Nigeria, African Journal of Pure and Applied Chemistry, 4(3): 22 - 30.

Yin, P., Wang, Z., Qu, R., Liu, X., Zhag, J and Xu, Q. (2012) Biosorption of Heavy Metal Ions onto Agricultural Residues Buckwheat Hulls Functionalized with 1-Hydroxylethylidenediphosporic Acid. J. Agric. Food Chem. 60(47): 11664-11674.